

The Native F_0F_1 -Inhibitor Protein Complex from Beef Heart Mitochondria and its Reconstitution in Liposomes

Edgar Vázquez-Contreras,¹ Nora Vázquez-Laslop,¹ and Georges Dreyfus^{1,2}

Received April 25, 1994; accepted June 22, 1994

A functional F_0F_1 ATP synthase that contains the endogenous inhibitor protein (F_0F_1I) was isolated by the use of two combined techniques [Adolfsen, R., McClung, J.A., and Moudrianakis, E. N. (1975). *Biochemistry* **14**, 1727–1735; Dreyfus, G., Celis, H., and Ramirez, J. (1984). *Anal. Biochem.* **142**, 215–220]. The preparation is composed of 18 subunits as judged by SDS-PAGE. A steady-state kinetic analysis of the latent ATP synthase complex at various concentrations of ATP showed a V_{\max} of $1.28 \mu\text{mol min}^{-1} \text{mg}^{-1}$, whereas the V_{\max} of the complex without the inhibitor was $8.3 \mu\text{mol min}^{-1} \text{mg}^{-1}$. In contrast, the K_m for Mg-ATP of F_0F_1I was $148 \mu\text{M}$, comparable to the K_m value of $142 \mu\text{M}$ of the F_0F_1 complex devoid of IF_1 . The hydrolytic activity of the F_0F_1I increased severalfold by incubation at 60°C at pH 6.8, reaching a maximal ATPase activity of $9.5 \mu\text{mol min}^{-1} \text{mg}^{-1}$; at pH 9.0 a rapid increase in the specific activity of hydrolysis was followed by a sharp drop in activity. The latent ATP synthase was reconstituted into liposomes by means of a column filtration method. The proteoliposomes showed ATP-Pi exchange activity which responded to phosphate concentration and was sensitive to energy transfer inhibitors like oligomycin and the uncoupler *p*-trifluoromethoxyphenylhydrazine.

KEY WORDS: F_0F_1 ATP synthase; inhibitor protein; protein reconstitution.

INTRODUCTION

The ATP synthases from cytoplasmic, thylakoidal, and mitochondrial inner membranes synthesize ATP coupled to an electrochemical gradient of protons generated by the photosynthetic or respiratory electron transport chain. These enzymes vary in complexity; nevertheless, their basic structure is conserved among a wide variety of phylogenetically diverse species. In all cases, the enzyme is arranged in two domains (for review see Futai *et al.*, 1989 and Pedersen and Amzel, 1993). The catalytic portion of the enzyme (F_1), consists of five subunits α , β , γ , δ , ϵ , which can be detached from membranes as a soluble ATPase. The integral membrane sector, F_0 , consists

of three subunits *a*, *b*, and *c* and functions as a proton pathway in ATP synthesis and hydrolysis. The ATP synthase complex catalyzes the hydrolysis and synthesis of ATP coupled to proton translocation and is thought to be regulated by an endogenous polypeptide referred to as the natural inhibitor protein. (Pullman and Monroy, 1963; for reviews, see Pedersen *et al.*, 1981, Gautheron and Godinot, 1988, and Harris and Das, 1991). The interest in obtaining an ATP synthase preparation that contains a functional inhibitor protein is that this will enable us to ask direct questions on the role of IF_1 on energy conservation and synthesis of ATP.

Here we have made use of two combined techniques (Adolfsen *et al.*, 1975 and Dreyfus *et al.*, 1984) to obtain a latent ATP synthase-inhibitor protein complex (F_0F_1I) with catalytic properties similar to the latent F_1 ATPase previously characterized by Feinstein and Moudrianakis (1984). The presence of the natural inhibitor protein was demonstrated by

¹ Departamento de Bioenergética, Instituto de Fisiología Celular, Universidad Nacional Autónoma de México, Apartado Postal 70-243, México D.F. 04510, Mexico.

² Corresponding author, email: gdreyfus@ifcsunl.ifsiol.unam.mx.

kinetic and by thermal and ion strength-induced increase in ATPase activity. The properties of this preparation contrast to those of ATP synthase, devoid of the regulating peptide, prepared as described by Dreyfus *et al.* (1984). The native ATP synthase complex reported here was reconstituted into phospholipid vesicles; the reconstituted enzyme carried out ATP-Pi exchange. This activity responds to inorganic phosphate and bovine serum albumin and is highly sensitive to the uncoupler *p*-trifluoromethoxyphenylhydrazone (FCCP) and oligomycin.

MATERIALS AND METHODS

Preparation of Submitochondrial Particles

Bovine heart mitochondria were prepared as described by Löw and Vallin (1963); "heavy mitochondria" were stored at -75°C until used. Submitochondrial particles devoid of inhibitor protein (IF₁) were prepared by resuspending mitochondria in a buffer containing 250 mM KCl, 75 mM sucrose, 30 mM Tris-SO₄ (pH 8.0), 2 mM EDTA, and processed as described by Klein *et al.* (1982). Submitochondrial particles with a high content of inhibitor protein (Mg-ATP particles) were prepared by resuspending mitochondria in a buffer containing 250 mM sucrose, 6 mM ATP, and 6 mM Mg-acetate and processed according to Lee and Ernster (1967). Submitochondrial particles were stored in liquid nitrogen until used.

Isolation and Reconstitution of the Mitochondrial F₀F₁ ATP Synthase

Isolation Procedure

In this work two different F₀F₁ ATP synthase preparations were isolated. F₀F₁ ATP synthase devoid of IF₁ and a latent form containing the endogenous inhibitor protein were obtained according to the method of Dreyfus *et al.* (1984), except that the solubilization of the F₀F₁ ATP synthase was performed with the nonionic detergent *n*-dodecyl- β -D-maltoside. For the isolation of the latent F₀F₁ ATP synthase, 13.5 mg of Mg-ATP submitochondrial particles were diluted to a concentration of 3 mg/ml in a buffer containing 150 mM sucrose, 20 mM Mes-Tris (pH 6.8), and 100 mM ADP. A volume of *n*-dodecyl- β -D-maltoside was added to the

submitochondrial particles suspension at a final concentration of 3.0 mg of detergent/mg of protein and incubated for 30 min at 4°C . The mixture was centrifuged 60 min at $100,000 \times g$. The pellet was resuspended in the same volume of the above-mentioned buffer and lauryl maltoside added at the same concentration. The supernatants were pooled and applied to a Sepharose-AH column (1 \times 5 cm) equilibrated with a buffer containing 20 mM Mes-Tris (pH 6.8), 2 mM EDTA, 10 mM sodium cholate, 5 mM ADP, and 3 mg/ml aroclorin; fractions of 5 ml were collected. The column was washed with 40 ml of the same buffer before eluting the retained F₀F₁ ATP synthase. Elution was carried out by the addition of 12.5 ml of the same buffer supplemented with 1 M KCl. The fractions were tested for ATPase activity and protein concentration, and concentrated by the addition of one volume of saturated ammonium sulfate (pH 6.8). After incubation at 4°C for 30 min, the sample was centrifuged 15 min at $20,500 \times g$. The pellet was resuspended in a buffer containing 250 mM sucrose, 20 mM Mes-Tris (pH 6.8), 2 mM EDTA, 2 mM sodium cholate, 100 μM ADP, and 30 mg/ml aroclorin and stored in liquid nitrogen until used.

Reconstitution into Liposomes

The ATP synthase complex was reconstituted using a variation of the elution-centrifugation method described by Garrett and Penefsky (1975) as modified by Vásquez-Contreras *et al.* (unpublished results). Freshly thawed (100–150 μg) F₀F₁ complex was applied to a 1-ml syringe containing Sephadex G50 fine previously equilibrated with Mes-Tris 20 mM (pH 6.8) and 2 mM EDTA. The column was centrifuged for 1 min at 3000 rpm in a Sorvall RT 6000B bench top centrifuge. Protein yield was ca. 90%.

Hydrolysis of ATP

ATP hydrolysis of the F₀F₁ and F₀F₁I ATP synthase (10–20 μg) was measured in 350 μl of a medium containing 100 mM Tris-SO₄ (pH 7.8), 3 mM ATP, 5 mM MgCl₂, 4 mM phosphoenolpyruvate, and 32 μg of pyruvate kinase. Incubation was carried out at 30°C for 10 min and the reaction was stopped by the addition of 100 μl trichloroacetic acid (6% final concentration). The amount of inorganic phosphate liberated was determined according to Fiske and SubbaRow (1925) or to

Lanzetta (1979) for a higher sensitivity. ATP hydrolysis was also determined spectrophotometrically as described by Pullman *et al.* (1960).

ATP-Pi Exchange Reaction

The proteoliposomes (100 μ g) of protein were diluted to 300 μ l in 250 mM sucrose. The reaction was started by the addition of 300 μ l of a medium containing 50 mM Tris acetate (pH 7.5), 2 or 15 mM H_3PO_4 (pH 7.2 with Tris base), 5 mM ADP, 3 mM ATP, 10 mM $MgSO_4$, 250 mM sucrose, and ^{32}P i (10⁷ cpm); in some experiments 1 mg/ml of bovine serum albumin was included. The reaction was incubated for 30 min at 30°C and stopped by the addition of TCA to 6% final concentration. The determination of ^{32}P i incorporated into ATP was performed as described by Dreyfus (1985).

Other Procedures

Immunoblotting was carried out after separating proteins by SDS-PAGE, and the proteins were electrophoretically transferred to nitrocellulose for 5 h (Towbin *et al.*, 1979). The sheets were blocked with 3% gelatin in H_2O for 1 h. Anti-IF₁ antiserum prepared as described by Dreyfus *et al.* (1981) was used at a 1:1000 dilution 1% gelatin in 100 mM Tris-HCl (pH 7.5), 0.9% NaCl, and 0.1% (vol/vol) Tween 20. The nitrocellulose sheets were then processed by using the Vectastain ABC-AP kit (Vector Laboratories, Burlingame, California).

Polyacrylamide gel electrophoresis was carried out in a gradient (10–23%) in the presence of 0.1% sodium dodecyl sulfate according to Laemmli (1970). The gels were scanned in an Ultrosan XL, LKB Bromma, Sweden using a Gel Scan Program from Pharmacia.

Protein determinations of the membrane fractions were measured as reported by Gornall *et al.* (1949) and according to Smith *et al.* (1985) for the soluble material. Bovine serum albumin was used as standard with both methods. Sepharose-AH, Sephadex G50 fine, phospholipids (phosphatidylcholine, type II-S from soybean), oligomycin, FCCP, dicyclohexylcarbodiimide (DCCD), and *n*-dodecyl- β -D-maltoside were from Sigma Chemical Co. Electron micrographs were taken with a Jeol microscope model 1200 EXII operated at 80 kV. Specimens were negatively stained with 2% (w/v) potassium phosphotungstate, pH 6.8.

RESULTS

Structural and Kinetic Characterization of F₀F₁ and F₀F₁I Preparations

Structural Characterization

A latent F₀F₁I ATP synthase was isolated from latent submitochondrial particles that had the ATP synthase in complex with the inhibitor protein (MgATP submitochondrial particles). The methodology employed was based on the ability of the F₁ ATPase to bind to Sepharose-AH (Tuena de Gómez-Puyou and Gómez-Puyou, 1977; Dreyfus *et al.*, 1984). The procedure developed renders a latent complex containing the natural ATPase inhibitor (IF₁). After solubilization with *n*-dodecyl- β -D-maltoside under acidic conditions in the presence of ADP, the enzyme was purified by Sepharose-AH chromatography. Under these conditions the interaction between F₁ and IF₁ is preserved. Figure 1 shows the electrophoretic pattern of F₀F₁I, the F₀F₁ ATP synthase, and IF₁. The ATP synthase complex is composed of 18 subunits which conform to F₁, the stalk, and F₀. It can be observed that IF₁ is present in the ATP synthase preparation (F₀F₁I). This was detected by Coomassie blue staining, but its presence was further confirmed by Western blot analysis using monospecific conventional antibodies (Fig. 2). It should also be noted that the ATP synthase complex isolated by this method contains two minor

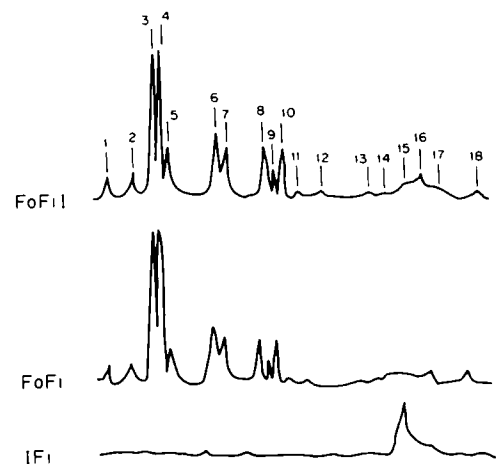


Fig. 1. Densitometric traces of F₀F₁-Inhibitor Protein, F₀F₁ complex, and IF₁ preparations. The gels were scanned in Ultrosan XL, LKB Bromma, Sweden. The proteins were separated by electrophoresis in the presence of sodium dodecyl sulfate on polyacrylamide gels as described in Materials and Methods.

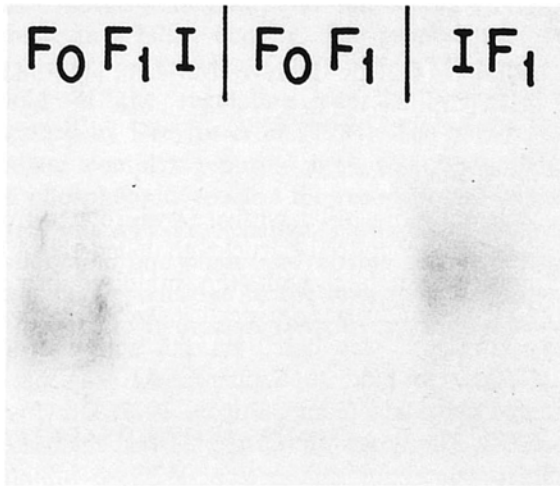


Fig. 2. Detection of IF₁ on the latent F₀F₁ ATP synthase by Western blot analysis. F₀F₁I as well as F₀F₁ preparations were analyzed using polyclonal anti-IF₁ antibodies. The analysis was performed as described under Materials and Methods. The gel was loaded with 20 μg of either F₀F₁I or F₀F₁ and 2 μg of IF₁.

Table I. Polypeptide Composition of the F₀F₁-Inhibitor Protein Complex^a

Band	Molecular weight	Subunit		
		F ₀	F ₁	Stalk
1	85,090			
2	69,800			
3	54,567		α	
4	50,998		β	
5	46,900		CK	
6	34,700			
7	30,300		γ	
8	24,500	"b"		
9	21,070			OSCP
10	19,700	"d"		
11	18,500	"a"		
12	15,000		δ	
13	12,400			F _B
14	11,900	"A6L"		
15	10,800			F ₆
16	10,100		IF ₁	
17	9,705	"c"		
18	8,400		"ε"	

^a The proteins were separated by electrophoresis in the presence of sodium dodecyl sulfate on polyacrylamide gel (10–23%) and stained with Coomassie Blue as described under Materials and Methods. CK corresponds to creatine kinase as reported by Lutter *et al.* (1993b).

components with molecular weights of 85,000 and 69,800 (see Table I). It also shows a band of molecular weight of 46,900 daltons that corresponds to creatine kinase as reported by Lutter *et al.* (1993b).

It is noteworthy that our preparations are essentially devoid of the adenine nucleotide carrier. This is of relevance since the carrier interferes with the reconstitution of the complex in isoelectric liposomes which was one of the aims of this work (*vide infra*).

Kinetic Characterization

In order to characterize the F₀F₁ ATP synthase in complex with IF₁, a steady-state kinetic analysis for Mg-ATP was carried out studying F₀F₁I, and F₀F₁ (Fig. 3). For the F₀F₁I Lineweaver–Burk plots showed that the complex had a V_{\max} of $1.3 \mu\text{mol min}^{-1} \text{mg}^{-1}$ and a K_m for Mg-ATP of $148 \mu\text{M}$ (closed circles). The F₀F₁ ATP synthase devoid of IF₁ shows a V_{\max} of $8.3 \mu\text{mol min}^{-1} \text{mg}^{-1}$ and a K_m for Mg-ATP of $142 \mu\text{M}$ (open circles). The ATPase activity of the latent F₀F₁I ATP synthase displayed a broad pH optima with a maximum at pH 9.0 (Fig. 4). The specific activity at pH 9.0 was severalfold higher than at pH 6.8, suggesting an activation of ATP hydrolysis due to a pH-dependent dissociation of the inhibitor protein (*vide infra*). It has been previously reported that the latent state of the mitochondrial F₁ ATPase in complex with the inhibitor protein can be abolished by incubation at 60°C (Feinstein and Moudrianakis, 1984; Beltrán *et al.* 1988). Hence we subjected the F₀F₁I ATP synthase, which exhibited low levels of ATPase activity, to a

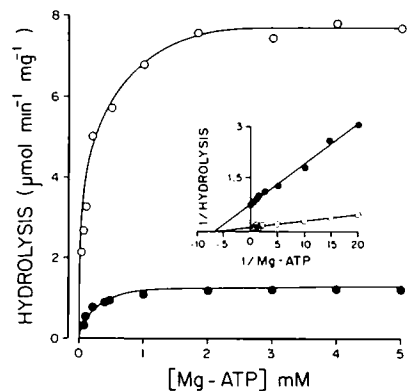


Fig. 3. Steady-state kinetic analysis for Mg-ATP of the latent and uncontrolled F₀F₁ ATP synthase. The F₀F₁I complex shows a V_{\max} value of $1.28 \mu\text{mol min}^{-1} \text{mg}^{-1}$ and a K_m for Mg-ATP of $148 \mu\text{M}$ (closed circles). On the other hand, the F₀F₁ ATP synthase devoid of IF₁ shows a V_{\max} of $8.33 \mu\text{mol min}^{-1} \text{mg}^{-1}$ and a K_m for Mg-ATP of $142 \mu\text{M}$ (open circles). ATPase activity was assayed as indicated under Materials and Methods.

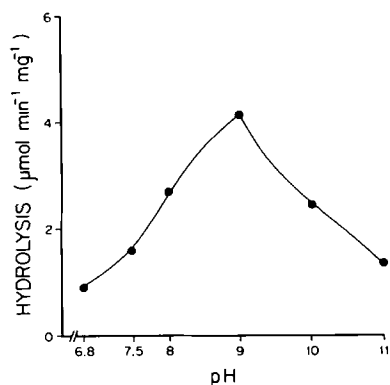


Fig. 4. Effect of pH on the ATPase activity of the F_0F_1I complex. ATP hydrolysis was determined as described under Materials and Methods. Samples were incubated for 10 min at the pH indicated and ATP hydrolysis was started by the addition of 350 μ l of reaction medium at 30°C.

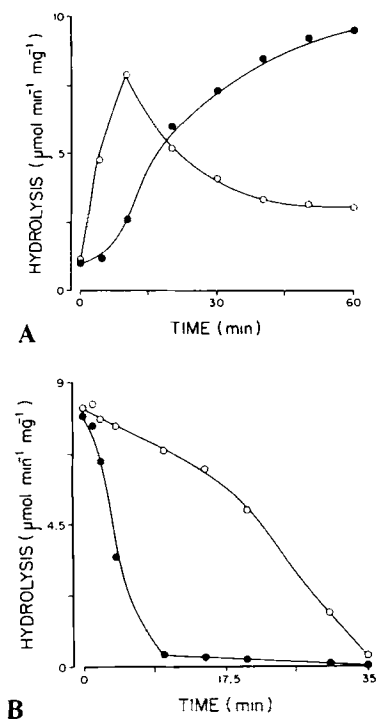


Fig. 5. (A) Effect of temperature (60°C) upon incubation of the F_0F_1I complex. pH 6.8 (closed circles) and pH 9.0 (open circles). The ATP synthase (10–20 μ g) was incubated at the temperature and pH indicated for 10 min. The samples were taken and assayed for ATP hydrolysis as described under Materials and Methods. (B) Effect of temperature (60°C) upon incubation of the F_0F_1 complex. pH 6.8 (closed circles) and pH 8.0 (open circles)

temperature of 60°C and a pH of 6.8. Under these conditions the ATPase activity gradually increased to a level 9.5 μ mol min⁻¹ mg⁻¹ (Fig. 5A). However, if the incubation at 60°C was carried out at pH 9.0, ATP hydrolysis increased rapidly to a maximum value of 7.9 μ mol min⁻¹ mg⁻¹; this was followed by a rapid drop in activity (Fig. 5A) that stabilized at about 3.0 μ mol min⁻¹ mg⁻¹. This is higher than the rate at time zero.

Interestingly, when the F_0F_1 ATP synthase was subjected to high temperature at a pH of 8.0, a decrease in the rate of ATP hydrolysis was observed as a function of time (open circles) in Fig. 5B. Almost complete inhibition is attained after 35 min. The decrease in hydrolytic activity was very sharp when the incubation was carried out at pH 6.8; inactivation was almost complete in 10 min of exposure to high temperature (Fig. 5B, closed circles).

The latent ATPase activity of the F_0F_1 -I ATP synthase could also be increased by incubation of the complex for 10 min with increasing concentrations of KCl. Complete activation was reached in the presence of 40 mM KCl (Fig. 6). It should be noted that the maximum activity reached was ca. 41% of the value obtained after 60 min at 60°C. The F_0F_1 -ATP synthase complex devoid of IF_1 displayed a high hydrolytic activity insensitive to potassium chloride (inset, Fig. 6).

Reconstitution of the ATP Synthase Complex in Liposomes and the ATP-Pi Exchange Reaction

The ATP synthase complex was reconstituted in liposomes (see Fig. 7) using the column filtration

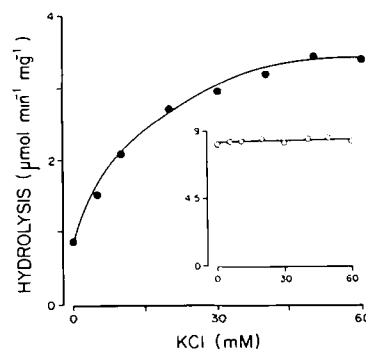


Fig. 6. Effect of increasing ionic strength on the ATPase activity of the F_0F_1I (closed circles) and F_0F_1 complex (open circles). Latent or active F_0F_1 ATP synthase was incubated in 100 μ l of buffer containing the indicated amounts of KCl. The hydrolysis of ATP was started by the addition of 250 μ l of reaction medium at 30°C as described under Materials and Methods.

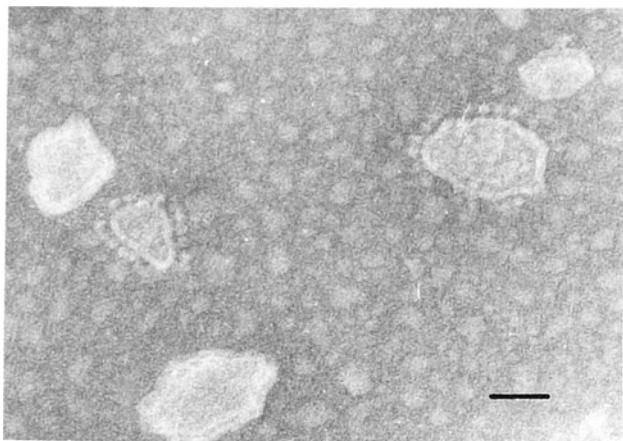


Fig. 7. Proteoliposomes of the reconstituted F_0F_1I complex. Bar corresponds to 50 nm.

method described under Materials and Methods. The ATP-Pi exchange reaction was assayed in media of various compositions in order to determine the conditions for maximal exchange (see Table II). Table II shows the exchange activity using two different concentrations of inorganic phosphate with and without bovine serum albumin (BSA). It can be noted that the exchange reaction carried out by the proteoliposomes containing the F_0F_1I complex responded to an increase in the concentration of phosphate, i.e., the activity increased from 2.5 with 2 mM H_3PO_4 to 6.4 $nmol\ min^{-1}\ mg^{-1}$ with 15 mM phosphate. In the presence of BSA and low phosphate concentration, the exchange rate was 4.87, and with the higher phosphate concentration, the rate was

Table II. ATP- ^{32}P i Exchange Reaction of the F_0F_1I and F_0F_1 Complex^a

Proteoliposomes	Additions		ATP-Pi exchange ($nmol\ min^{-1}\ mg^{-1}$)		
	H_3PO_4 (mM)	BSA (mg/ml)	Control	+FCCP	+Oligomycin
F_0F_1I	2	—	2.5	0	0
	2	1	4.9	0	0
	15	—	6.4	0	0.1
	15	1	11.0	0	0.3
F_0F_1	2	—	0.2	0	0
	2	1	0.7	0	0
	15	—	1.8	0	0
	15	1	4.1	0	0

^a The ATP-Pi exchange reaction was assayed under various conditions in order to determine the optimal rate of the exchange reaction.

11.0 $nmol\ min^{-1}\ mg^{-1}$. The ATP-Pi exchange reaction was inhibited by nearly 100% oligomycin or FCCP (0.8 μg and 1 μM final concentration, respectively). Interestingly, the ATP-Pi exchange activity of the F_0F_1 ATP synthase was significantly lower than of the latent F_0F_1I regardless of the fact that the hydrolytic activity of the former preparation is higher than with a consequent higher capacity to build up an ATP-dependent electrochemical gradient. The exchange reaction catalyzed by F_0F_1 with or without IF_1 was almost completely inhibited by FCCP and oligomycin. This illustrates that the coupling between F_1 and the proton channel F_0 is well preserved after the column filtration treatment.

DISCUSSION

The isolation technique used in the present study has allowed us to obtain an ATP synthase that is functional and under control of the endogenous inhibitor protein. The complex is composed of 18 polypeptides. Its composition is similar to the preparations of Galante *et al.* (1979), Dreyfus *et al.* (1984), Laird *et al.* (1986), and Walker *et al.* (1991). It should be noted that the subunit composition of the ATPase complex varies slightly among the preparations reported in the literature; this could be related to the various methodologies that have been used. Of the above-mentioned preparations only the one reported by Walker *et al.* (1991) contains endogenous IF_1 . Therefore the true subunit composition of a functional F_0F_1 ATP synthase is still under debate. The preparation presented in this study contains the natural inhibitor protein IF_1 and is devoid of the adenine nucleotide carrier, in contrast to the preparation of Serrano *et al.* (1976). The presence of IF_1 as a component that copurifies with the complex was evidenced by Western blot analysis using specific polyclonal antibodies. It is important to note that an ATP synthase preparation such as the one presented in this study will be of great use in the study of the regulating mechanisms in the native enzyme.

The latent F_0F_1I responds to heat activation in a manner similar to the latent and soluble F_1I ATPase described by Feinstein and Moudrianakis (1984) and the controlled submitochondrial particles of Beltrán *et al.* (1988). The kinetic constants determined for the latent complex show a low rate of ATP hydrolysis and an affinity for ATP that is in accordance with the values reported for the reconstituted ATP synthase

complex studied by Stigall *et al.* (1978) and Laird *et al.* (1986) which are devoid of IF₁. It should be noted that the affinity for ATP was not modified by the presence or absence of IF₁. Furthermore, we performed determinations of the affinity for Mg-ATP under conditions of saturating amounts of IF₁ (data not shown); this did not modify the K_m value for Mg-ATP. These results are of importance in the light of the fact that the F₀F₁ ATP synthase exhibits a higher hydrolytic activity and the ATP-Pi exchange activity is significantly lower than in the case of the latent enzyme. The structural integrity of the reconstituted complex was demonstrated by the nearly complete inhibition of the hydrolytic activity by oligomycin. On the other hand, at pH 9.0 the latent F₀F₁I ATP synthase showed a specific activity which was severalfold higher than at pH 7.4, suggesting that at alkaline pH IF₁ is released from the enzyme. This is in accordance with the notion of Milgrom (1991) that an inactive form of IF₁ is formed at high pH, through the rapid conversion of an active acidic form to an inactive basic form of IF₁.

Many unsuccessful efforts to reconstitute the latent F₀F₁I and activated ATP synthase complex F₀F₁ by dialysis were carried out during the development of this study; the enzyme was always inactivated after several hours of dialysis. This result may be related to the detergent used to solubilize the enzyme. We have previously reported that the ATP synthase complex solubilized from the membrane with lauryl dimethyl amino oxide (LDAO) (Dreyfus *et al.* 1984) could be reconstituted by dialysis. We have now developed a reconstitution technique in which a central step is the passage of the complex through centrifugation columns (Vázquez-Contreras *et al.*, unpublished results). The resulting ATP synthase complex exhibits hydrolytic activity and a high sensitivity to oligomycin, and can be readily incorporated into liposomes with or without IF₁. This allowed us to determine the extent of ATP-Pi exchange experiments in an integral F₀F₁ with and without endogenous IF₁ complex. These results are in contrast to those of Lutter *et al.* (1993a) who were not able to demonstrate either ATP-Pi exchange or ATP synthesis in a similar ATP synthase preparation. Both preparations catalyzed an ATP-Pi exchange reaction that was sensitive to FCCP and oligomycin. This reaction increased as the concentration of phosphate was increased, and bovine serum albumin increased phosphate uptake. The latter was probably due to a scavenger effect of

detergent molecules that hinder the ATP-Pi exchange reaction. The latent F₀F₁I ATP synthase has a higher specific exchange activity than the ATPase complex devoid of IF₁. This is particularly relevant, since the hydrolytic activity of F₀F₁I ATP synthase is severalfold lower than that of complex without IF₁. Hence the controlled ATP synthase shows a tighter coupling between hydrolytic and synthetic reactions. We have compared the ATP-Pi exchange rates displayed by the F₀F₁I and F₀F₁ ATP synthase preparations with those reported by Serrano *et al.* (1976), Stigall *et al.* (1978), and Dreyfus (1985). From previously reported data we cannot conclude if the absolute rate value for ATP-Pi exchange should be compared. A more meaningful term will be the ratio of ATP hydrolyzed to Pi uptake as this reflects the true value of energy conservation. This suggests that IF₁ would not only regulate catalysis by inhibiting ATP hydrolysis but would be acting as a real coupling device that increases the efficiency of energy transduction. In fact, Guerrieri *et al.* (1987) have shown that IF₁ not only inhibits ATP hydrolysis, but also inhibits the proton conductivity of the complex. These authors suggest that the inhibitor protein could be involved in the gate of the H⁺-ATPase.

There has been some discussion whether IF₁ is dissociated from the ATP synthase under energized conditions (see the review by Pedersen *et al.*, 1981; and Harris and Das, 1991). However, recent data showed convincingly that IF₁ remains in the ATP synthase under such conditions (López-Mediavilla *et al.*, 1993), which is in agreement with our previous observations (Dreyfus *et al.*, 1981).

The ATP synthase preparation presented in this study retains its controlled hydrolytic activity, due to the presence of endogenous IF₁. This provides the possibility of carrying out detailed studies on the regulation of the catalytic activity of the native latent ATP synthase complex and on the possible role that IF₁ might exert on energy conservation.

ACKNOWLEDGMENTS

This work was partially supported by a grant (0617-N9110) from the Consejo Nacional de Ciencia y Tecnología (CONACyT). We also like to thank Dr Diego González-Halphen and Prof. Armando Gómez-Puyou for helpful discussions and Bertha González and Ernesto Maldonado for revising the manuscript. We also like to thank Teresa Ballado

for expert technical assistance, Jorge Sepulveda for carrying out the electron microscopic analysis, and José García Trejo for valuable help with the exchange reactions.

REFERENCES

- Adolfson, R., McClung, J. A., and Moudrianakis, E. N. (1975). *Biochemistry* **14**, 1727-1735.
- Beltrán, C., Gómez-Puyou, A., and Tuena de Gómez-Puyou, M. (1988). *Biochem. Biophys. Res. Commun.* **152**, 867-873.
- Dreyfus, G. (1985). *J. Biol. Chem.* **260**, 12112-12117.
- Dreyfus, G., Gómez-Puyou, A., and Gómez-Puyou, M. T. (1981). *Biochem. Biophys. Res. Commun.* **100**, 400-406.
- Dreyfus, G., Celis, H., and Ramirez, J. (1984). *Anal. Biochem.* **142**, 215-220.
- Feinstein, D.L., and Moudrianakis, E. N. (1984). *J. Biol. Chem.* **259**, 4230-4236.
- Fiske, C. H., and Subbarow, Y. (1925). *J. Biol. Chem.* **66**, 375-400.
- Futai, M., Noumi, T., and Maeda, M. (1989). *Annu. Rev. Biochem.* **58**, 111-136.
- Galante, Y. M., Wong, S. Y., and Hatefi, Y. (1979). *J. Biol. Chem.* **254**, 12372-12378.
- Garrett, N. E., and Penefsky, H. S. (1975). *J. Biol. Chem.* **250**, 6640-6647.
- Gautheron, D. C., and Godinot, C. (1988). *J. Bioenerg. Biomembr.* **20**, 451-468.
- Gornall, A. G., Bardawill, C. J., and David, M. M. (1949). *J. Biol. Chem.* **177**, 751-766.
- Guerrieri, F., Scarfo, R., Zanotti, F., Che, Y-W., and Papa, S. (1987). *FEBS Lett.* **213**, 67-72.
- Harris, D. A., and Das, A. M. (1991). *Biochem. J.* **280**, 561-573.
- Klein, G., Satre, M., Zaccari, G., and Vignais, P. (1982). *Biochim. Biophys. Acta* **681**, 226-232.
- Laemmli, U. K. (1970). *Nature (London)* **227**, 680-685.
- Laird, D. M., Eble, K. S., and Cunningham, C. C. (1986). *J. Biol. Chem.* **261**, 14844-14850.
- Lanzetta, A. (1979). *Anal. Biochem.* **100**, 95-97.
- Lec, C. P., and Ernster, L. (1967). *Methods Enzymol.* **10**, 543-548.
- López-Mediavilla, C., Vigny, H., and Godinot, C. (1993). *Eur. J. Biochem.* **215**, 487-496.
- Löw, H., and Vallin, I. (1963). *Biochim. Biophys. Acta* **69**, 361-374.
- Lutter, R., Sarastre, M., Van Walraven, H., Runswick, M. J., Fincl, M., Deatherage, J. F., and Walker, J. E. (1993a). *Biochem. J.* **295**, 799-806.
- Lutter, R., Abrahams, J. P., van Raaij, M. J., Todd, R. J., Lundqvist, T., Buchanan, S. K., Leslie, A. G. W., and Walker, J. E. (1993b). *J. Mol. Biol.* **229**, 787-790.
- Milgrom, Y. (1991). *Eur. J. Biochem.* **200**, 789-795.
- Pedersen, P. L., and Amzel, L. M. (1993). *J. Biol. Chem.* **268**, 9937-9940.
- Pedersen, P. L., Schwerzmann, K., and Cintron, N. (1981). *Curr. Top. Bioenerg.* **11**, 149-199.
- Pullman, M. E., and Monroy, G. C. (1963). *J. Biol. Chem.* **238**, 3762-3769.
- Pullman, M. E., Penefsky, H. S., Datta, A., and Racker, E. (1960). *J. Biol. Chem.* **253**, 3322-3329.
- Serrano, R., Kanner, B. I., and Raker, E. (1976). *J. Biol. Chem.* **251**, 4253-4261.
- Smith, P. K., Krohn, R. I., Hermanson, G. T., Mallia, A. K., Gartner, F. H., Provenzano, M. D., Fujimoto, E. K., Goeke, N. M., Olson, B. J., and Klenk, D. (1985). *Anal. Biochem.* **150**, 76-85.
- Stigall, D., Galante, Y. M., and Hatefi, Y. (1978). *J. Biol. Chem.* **253**, 956-964.
- Towbin, H., Staehelin, T., and Gordon, J. (1979). *Proc. Natl. Acad. Sci. USA* **76**, 4350-4354.
- Tuena de Gómez-Puyou, M., and Gómez-Puyou, A. (1977). *Arch. Biochem. Biophys.* **182**, 82-86.
- Walker, J. E., Lutter, R., Dupuis, A., and Runswick, M. J. (1991). *Biochemistry* **30**, 5369-5378.